

# EFFECTS OF DIFFERENT EDIBLE COATINGS IN PHYSICAL, CHEMICAL AND MICROBIOLOGICAL CHARACTERISTICS OF MINIMALLY PROCESSED PEACHES (*PRUNUS PERSICA L. BATSCH*)

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## ABSTRACT

The present work aimed at evaluating the effects of tara gum, xanthan and sodium alginate as edible coatings in physical, chemical and microbiological characteristics of minimally processed peaches. Peaches from Granada cultivar were peeled, cut into cubes and immersed in the solutions with the respective treatment. Four treatments were used. The minimally processed peaches were packed in PET packaging and stored at  $4 \pm 1$  C for 12 days. Physical and physicochemical analyses were performed, including mass loss, firmness (N), color ( $L^*$ ,  $a^*$ ,  $b^*$  and Hue angle), pH, soluble solids ( $^{\circ}$ Brix), and microbiological evaluations of psychotrophic microorganisms and mold and yeast. The treatment with tara gum associated to citric and ascorbic acids and sodium chloride presented the best results. Tara gum allowed a reduction of mass loss, firmness maintenance, lower color alteration and lower growth of molds and yeasts. Therefore, this gum presents potential for application as edible coating.

## PRACTICAL APPLICATIONS

Fresh-cut peaches are appreciated worldwide for their exotic flavor and nutritional composition. However, their shelf life is limited by changes in color, texture, appearance and microbial growth. An alternative to try to minimize the losses and increase the shelf life of this product is the use of edible coatings, maintaining the quality parameters of fresh fruit. Tara gum is a *cheap gum* that has not been used as coating on minimally processed fruit, and in this work, this gum showed significant results in the quality of fresh-cut peaches for a longer time. It can be considered useful to the minimal processing industry and used as an alternative to fast food and other ready-to-eat products, attending the demand for healthy and convenient foodstuffs.

## INTRODUCTION

Peach (*Prunus persica* L. Batsch) is a climacteric fruit originating from Asia. It presents expressive commercial production worldwide mainly in regions of temperate climate (Scorza 2005). Brazil is the 15th largest world producer, despite its favorable geographic position and its edaphic climatic conditions. Among the producer states, Rio Grande

do Sul stands out. It is responsible for around 65% of national production (Sebrae 2012). Peach peculiarities of flavor and aroma result from a balance of sugars, organic acids, phenolic compounds, carotenoids and volatile compounds, making it an appreciated fruit of great commercial importance (Gil *et al.* 2002; Versari *et al.* 2002).

Minimally processed fruits offer a convenient and practical product to the consumer, with quality and freshness

similar to the *in natura* product. According to Della Cruz (2004), this kind of product provides higher yield, good quality, reduces accidents in kitchens and requires less refrigerated space. Besides the advantages of convenience and quality offered by minimally processed fruits, the possibility of processing them in the producing regions provides a new option to the producers as it adds value to products and is suitable for micro and small family business.

An alternative to reduce the effects caused by the minimal process is the use of edible coatings. They form a thin layer of edible material on the product surface. They are used to inhibit the migration of moisture, oxygen, carbon dioxide, aroma and lipids; introducing additives such as antioxidants and antimicrobials, thereby improving the characteristics, mechanical integrity and handling of food products (Krochta and Mulder-Johnston 1997). However, edible coatings have not been used in the preservation of minimally processed peaches, only in the preservation of the whole fruit as in studies carried out by Oliveira and Cereda (2003), Togrul and Arslan (2004) and Maftoonazad *et al.* (2008).

Tara gum is extracted from *Caesalpinia spinosa* endosperm seeds. It is a galactomannan, a neutral polysaccharide used as a thickening agent in foods and, unlike the alginates and gelatin gum, it does not form gel on its own (Sittikijyothin *et al.* 2005). *Caesalpinia spinosa* is native from Bolivia, Peru and Northern Chile; but it is also found in Ecuador, Colombia, Venezuela and Cuba. It is believed that Peru is the largest tara gum exporter. Tara gum presents a low relative cost, offering a great economic incentive to its industrial exploitation (Fsanj 2006). So far, it has not been evaluated as edible coating of minimally processed food.

Xanthan gum is a polysaccharide synthesized by a phytopathogenic bacterium from *Xanthomonas* species (Sutherland 1993). Xanthan gum is used in foods as thickener and stabilizer; however, it does not form gel on its own, only in synergism with galactomannans (García-Ochoa *et al.* 2000).

Alginates are polysaccharides most commonly used in edible coatings. Sodium alginate is a polysaccharide of natural origin derived mainly from brown seaweed (Çaykara *et al.* 2005).

Therefore, the aim of the present study was to evaluate the effects of tara gum, xanthan gum and sodium alginate as edible coatings on the microbiological, physical and physicochemical characteristics of minimally processed peaches.

## MATERIALS AND METHODS

### Material

Peaches (*P. persica* L. Batsch) from Granada cultivar were bought from grocery stores in Rio Grande City, with maturity level between 50% and 70%. The fruits were selected as per size, color and the absence of physiological defects. They

were transported in thermo boxes to the Laboratory of Food Technology of the School of Chemistry and Food of Federal University of Rio Grande, where the minimal processing was carried out.

### Preparation of Minimally Processed Peaches

The processing was carried out at approximately 10°C, with the utensils previously cleaned in a solution of organic chlorine (dichloroisocyanurate) at a concentration of 2 g/l. Peaches were selected, washed and sanitized in organic chlorine solution at a concentration of 2 g/l for 15 min, peeled and manually cut into cubes of approximately 2.5 × 2.5 cm using stainless steel knives. Then the pieces were rinsed with chlorinated water (0.2 g/l) to eliminate the cell juice spilled.

### Preparation and Coating Application

Xanthan gum (Shandong Fufeng), tara gum (Silva Extract Srl) and sodium alginate (Kimica) were slowly dissolved in water under constant agitation for about 1 h. Then xanthan gum was heated at 60°C for 20 min (Xuewu *et al.* 1996), tara gum at 80°C for 30 min (Sittikijyothin *et al.* 2007) and sodium alginate at 70°C for 30 min, then cooled at 15°C (Rojas-Graü *et al.* 2007). After cooling the solutions, calcium chloride, citric and ascorbic acids and glycerol were added under agitation.

The coatings were prepared in aqueous solution: treatment 1 – control (cut peach); treatment 2 – alginate (2% w/v), ascorbic acid (1% w/v), citric acid (0.25% w/v), CaCl<sub>2</sub> (1% w/v) and glycerol (1% v/v); treatment 3 – xanthan gum (0.5% w/v), ascorbic acid (1% w/v), citric acid (0.25% w/v), CaCl<sub>2</sub> (1% w/v) and glycerol (1% v/v); treatment 4 – tara gum (0.5% w/v), ascorbic acid (1% w/v), citric acid (0.25% w/v), CaCl<sub>2</sub> (1% w/v) and glycerol (1% v/v).

The cubes were fully submerged into the solution for 1 min and drained by the use of nylon nets for 2 to 3 min to eliminate the excess of solution. The cubes were dried using a fan in a refrigerated environment for 30 min. Finally, the samples were packed in non-recycled polyethylene terephthalate – PET packaging with a lid, with the same number of pieces per packaging, and stored at 4°C for 12 days. The physical, physicochemical and microbiological tests were performed in triplicate on the day of processing the samples (day 0) and after 1, 3, 5, 7, 9 and 12 days of storage.

### Physical Analysis

**Weight Loss.** The weight loss was obtained by taking the difference between the initial weight of the minimally processed peach and that obtained at the end of each storage time, according to the formula: (%) Weight loss = [(initial

mass – final mass)/(initial mass) × 100. The results were expressed as percentage of weight loss.

### Texture Analysis

The measurements of peach cube firmness were determined by using a texture analyzer Model TA-XT2 plus (Stable Micro Systems, Surrey, England). A cylindrical probe in the pre-test speed of 4 mm/s, post test of 8 mm/s, test of 2 mm/s and penetration depth of 5 mm was used. The results were expressed in Newton (N).

**Color.** The color analysis was evaluated by using a Minolta colorimeter, model Chroma Meter CR400. The parameters  $L^*$  (0 [black] to 100 [white]),  $a^*$  (green chromaticity [–60] to red [+60]) and  $b^*$  (blue chromaticity [–60] to yellow [+60]) and *Hue* angle (showing the location of a color diagram, where the angle 0° represents pure red, 90° represents pure yellow, 180° pure green and 270° pure blue). A standard white calibration plate was used (Djioua *et al.* 2009).

**pH.** The pH was determined using the method described by AOAC (2000). pH was measured using a digital pH meter (Model PA 200, Marconi Instruments, Inc., Piracicaba, SP). A suspension of 20 g of sample in 100 ml of distilled water was prepared, thus measuring the pH with the aid of a pH meter.

**Total Soluble Solids.** The content of total soluble solids was determined in a bench-type refractometer Abbé, with temperature correction to 20°C. The results were expressed in °Brix (AOAC 2000).

### Microbiological Analysis

The microbiological tests were carried out to detect the presence of psychrotrophic spp., and yeasts and molds, following the method described in APHA (2001).

### Statistical Analysis

The physical and physicochemical evaluations were submitted to variance analysis and comparison of the mean was done by Tukey's test with a significance level of 5%, using Statistica 7.0 software (StatSoft, Inc., Tulsa, OK).

## RESULTS AND DISCUSSION

### Physical and Physicochemical Analyses

Table 1 shows the values of mass loss of minimally processed peaches with different coatings, stored at  $4 \pm 1$ °C for 12 days.

There was an increase in the mass loss of minimally processed peaches during storage in all treatments. However, the loss was significantly lower in coated peaches than in the control (T1). This treatment presented the highest mass loss, 34.1% in the 12 days of storage. Treatment T4 presented the lowest loss (7.60%), whereas treatment T3 presented the highest mass loss among the coated samples (10.87%). According to Raybaudi-Massilia *et al.* (2007) and Villalobos-Carvajal *et al.* (2009), edible coatings can reduce mass loss because they help to decrease water loss from minimally processed products. However, in this work, tara gum presented the best results in relation of mass loss.

There are no reports of studies that show mass loss in minimally processed peaches, only using the whole fruit. Jacometti *et al.* (2003) evaluated mass loss of whole peaches coated with gelatin stored at 10°C for 16 days; however, they did not observe significant difference in mass loss of coated fruits with gelatin and the uncoated control. Oliveira and Cereda (2003) showed that the mass loss of whole peaches stored at 27°C during 12 days is dependent on the coating applied. Treatment with commercial wax and micro emulsion allowed less mass loss when compared to control treatment and with cassava starch. Maftoonzad *et al.* (2008) evaluated reduction of loss of moisture of

Days	Treatments			
	T1	T2	T3	T4
0	0.0 <sup>9A</sup>	0.0 <sup>9A</sup>	0.0 <sup>9A</sup>	0.0 <sup>9A</sup>
1	1.27 ± 0.02 <sup>1A</sup>	0.73 ± 0.01 <sup>1C</sup>	1.06 ± 0.02 <sup>1B</sup>	0.83 ± 0.01 <sup>1B</sup>
3	2.22 ± 0.01 <sup>1C</sup>	3.09 ± 0.01 <sup>1A</sup>	2.89 ± 0.01 <sup>1B</sup>	1.35 ± 0.02 <sup>1D</sup>
5	4.17 ± 0.03 <sup>1C</sup>	4.69 ± 0.02 <sup>1B</sup>	5.94 ± 0.01 <sup>1A</sup>	2.16 ± 0.01 <sup>1D</sup>
7	8.95 ± 0.01 <sup>1A</sup>	5.69 ± 0.02 <sup>1C</sup>	6.53 ± 0.01 <sup>1B</sup>	2.88 ± 0.01 <sup>1D</sup>
9	21.77 ± 0.01 <sup>1A</sup>	6.81 ± 0.01 <sup>1B</sup>	7.13 ± 0.02 <sup>1B</sup>	4.09 ± 0.02 <sup>1D</sup>
12	34.1 ± 0.03 <sup>1A</sup>	8.38 ± 0.02 <sup>1C</sup>	10.87 ± 0.03 <sup>1B</sup>	7.6 ± 0.02 <sup>1D</sup>

Means followed by the same letter in the column and capital letter in the row did not differ by Tukey's test ( $P < 0.05$ ). (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.

**TABLE 1.** MASS LOSS (%) OF MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS, STORED AT  $4 \pm 1$ °C FOR 12 DAYS

**TABLE 2.** FIRMNESS VALUES (N) OF MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS, STORED AT  $4 \pm 1$  C FOR 12 DAYS

Days	Treatments			
	T1	T2	T3	T4
0	14.70 $\pm$ 0.71 <sup>eC</sup>	14.90 $\pm$ 0.12 <sup>dC</sup>	17.76 $\pm$ 1.03 <sup>bB</sup>	20.13 $\pm$ 0.92 <sup>aA</sup>
1	14.71 $\pm$ 0.68 <sup>eC</sup>	14.92 $\pm$ 0.17 <sup>cdC</sup>	17.77 $\pm$ 1.05 <sup>bB</sup>	20.13 $\pm$ 0.86 <sup>aA</sup>
3	15.92 $\pm$ 0.21 <sup>dC</sup>	15.78 $\pm$ 0.25 <sup>bcC</sup>	18.65 $\pm$ 0.58 <sup>abB</sup>	20.87 $\pm$ 0.91 <sup>aA</sup>
5	17.28 $\pm$ 0.17 <sup>cC</sup>	16.43 $\pm$ 0.35 <sup>abC</sup>	19.09 $\pm$ 0.71 <sup>abB</sup>	21.19 $\pm$ 0.77 <sup>aA</sup>
7	18.50 $\pm$ 0.32 <sup>bB</sup>	16.72 $\pm$ 0.22 <sup>aC</sup>	19.52 $\pm$ 0.85 <sup>abB</sup>	21.43 $\pm$ 0.88 <sup>aA</sup>
9	18.75 $\pm$ 0.16 <sup>abC</sup>	16.47 $\pm$ 0.08 <sup>abD</sup>	19.81 $\pm$ 0.46 <sup>abB</sup>	21.23 $\pm$ 0.21 <sup>aA</sup>
12	19.87 $\pm$ 0.19 <sup>aB</sup>	16.04 $\pm$ 0.11 <sup>abC</sup>	20.75 $\pm$ 0.69 <sup>aA</sup>	21.07 $\pm$ 0.19 <sup>aA</sup>

Means followed by the same letter in the column and capital letter in the row did not differ by Tukey's test ( $P < 0.05$ ). (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%,  $\text{CaCl}_2$  1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%,  $\text{CaCl}_2$  1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%,  $\text{CaCl}_2$  1%, glycerol 1%.

whole peaches by the use of coatings based on methyl cellulose and sodium alginate in 24 days of storage at 15C. After 12 days of storage, the mass loss in the uncoated control peaches was three times higher than the peaches coated with methyl cellulose and two times with sodium alginate.

Table 2 shows firmness values ( $N$ ) of minimally processed peaches using different coatings, stored at  $4 \pm 1$  C for 12 days.

There was an increase of firmness in all treatments; however, in the treatment with tara gum (T4), the difference was not significant during the whole period of storage. The increase of firmness from the first to the last day of storage was 4.46%. The control treatment showed higher increase in firmness when comparing the first and the last day of storage, reaching 26%, followed by the treatment with xanthan, 14.4%, and alginate 7.1%. There was no connection between firmness increase and calcium chloride addition, since this increase was also observed in the control sample (T1), which did not receive calcium addition. However, there was a correlation between the values obtained in the mass loss with the increase in the firmness. This increase in firmness values may have been caused by water loss through evaporation and exudation, which promote drying of the surface tissues. According to Souza *et al.* (2005), this effect in minimally processed papaya is due to formation of a resistant superficial tissue due to higher moisture loss, making the pieces to become firmer during storage.

Chagas *et al.* (2008) evaluated firmness of minimally processed peaches treated with different concentrations of citric acid. According to their results, firmness was maintained in the first 9 days with a subsequent reduction. In the study carried out by Martins (2010), in which minimally processed peaches were treated with different concentrations of ascorbic acid and calcium chloride, there was maintenance of firmness in the samples treated with ascorbic acid and increase in the ones treated with calcium chloride in 12 days of storage at 3C. When different packaging materials were

evaluated by the authors, the maintenance of firmness occurred mainly with the use of polyethylene terephthalate packaging (PET), in relation to polyvinylchloride and polyolefin.

In a study by Costa (2010), the maintenance of firmness was observed in minimally processed peaches treated with ascorbic acid and sodium chloride and also in the control sample. In the other treatments (L-cysteine hydrochloride and calcium chloride; L-cysteine hydrochloride, ascorbic acid and calcium chloride), there was a firmness reduction tendency justified by the increase in the metabolic activity, enzymes and substrates decompartmentalization, which promote depolymerization of pectin. Oliveira and Cereda (2003) evaluated the firmness of whole peaches treated with different coatings including cassava starch, microemulsion based on cassava starch and commercial wax. In all treatments, a reduction of firmness values during storage was observed. These results agree with the ones obtained by Maftoonazad *et al.* (2008); however, according to them, the coated fruits showed a beneficial effect in firmness maintenance when compared with the control sample. Methyl cellulose was more efficient in retaining firmness than sodium alginate.

Table 3 shows values of lightness ( $L^*$ ) of minimally processed peaches using different coatings, stored at  $4 \pm 1$  C for 12 days.

There was a reduction in  $L^*$  values in all treatments, suggesting browning of minimally processed peaches. The lowest reduction in  $L^*$  values was observed using coatings T4 (6.42%) and T3 (10.40%), while for the control treatment, the reduction in  $L^*$  values was 17.20% and in T2 it was 14.45%.

The highest lightness loss in the control sample can be explained because the sample was not treated with the antioxidants ascorbic and citric acids. These are effective agents in the control of enzymatic browning, as they are capable of reducing quinones to phenols (Torralles *et al.* 2008). However, similar to this study, other works also show that

Days	Treatments			
	T1	T2	T3	T4
0	63.76 ± 0.24 <sup>aC</sup>	67.04 ± 0.16 <sup>aB</sup>	67.60 ± 0.05 <sup>aA</sup>	67.80 ± 0.27 <sup>aA</sup>
1	62.05 ± 0.97 <sup>bB</sup>	66.23 ± 0.07 <sup>abA</sup>	66.44 ± 0.14 <sup>bA</sup>	67.50 ± 0.21 <sup>aA</sup>
3	60.89 ± 0.67 <sup>cC</sup>	64.16 ± 0.65 <sup>bcB</sup>	65.34 ± 0.34 <sup>cA</sup>	65.36 ± 0.16 <sup>bA</sup>
5	60.72 ± 0.11 <sup>cC</sup>	62.68 ± 0.31 <sup>cdB</sup>	64.96 ± 0.14 <sup>dA</sup>	65.35 ± 0.15 <sup>bA</sup>
7	58.11 ± 0.28 <sup>dD</sup>	60.9 ± 0.21 <sup>deC</sup>	63.59 ± 0.08 <sup>eB</sup>	64.71 ± 0.27 <sup>bCA</sup>
9	56.57 ± 0.07 <sup>eD</sup>	59.72 ± 0.19 <sup>efC</sup>	62.53 ± 0.16 <sup>eB</sup>	64.18 ± 0.52 <sup>cdA</sup>
12	52.81 ± 0.37 <sup>fD</sup>	57.35 ± 0.26 <sup>fC</sup>	60.57 ± 0.07 <sup>fB</sup>	63.45 ± 0.18 <sup>dA</sup>

Means followed by the same letter in the column and capital letter in the line did not differ by Tukey's test ( $P < 0.05$ ). (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.

even when using antioxidant agents, the color of minimally processed peaches is partially altered.

Chagas *et al.* (2008) observed a decrease in the lightness values ( $L^*$ ) during storage of minimally processed peaches treated with different concentrations of citric acid. A study by Martins (2010) also observed reduction in the lightness values in 12 days of storage of minimally processed peaches submitted to treatments with ascorbic acid and calcium chloride and different packaging materials. Yet at the end of the storage period, there was no significant difference between the control and the different treatments. Costa (2010) observed a decline in lightness values in samples of minimally processed peaches treated with ascorbic acid and calcium chloride; L-cysteine hydrochloride and calcium chloride; L-cysteine hydrochloride, ascorbic acid and calcium chloride. However, the lowest values of  $L^*$  were observed in the control treatment.

Table 4 shows the values of  $a^*$  of minimally processed peaches using different coatings, stored at  $4 \pm 1$  C for 12 days.

Values of  $a^*$  in the different treatments oscillated during the period evaluated, tending to increase toward the end of the storage period. This increase indicates higher intensity of red color. The lowest increase in the values of  $a^*$  was

**TABLE 3.** LIGHTNESS VALUES ( $L^*$ ) OF MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS, STORED AT  $4 \pm 1$  C FOR 12 DAYS

observed using T4 (36.9%) and T3 (38.5%) coatings. In the control treatment, there was an increase of 48.2%. T2 resulted in an increase of 43.8%. Different results were found by Chagas *et al.* (2008). According to their results, values of  $a^*$  did not differ significantly during the 9 days of storage and neither among minimally processed peaches treated with citric acid (1% and 2%).

Table 5 shows values of  $b^*$  of minimally processed peaches using different coatings, stored at  $4 \pm 1$  C for 12 days. There were also fluctuations in the values of  $b^*$  during storage; however, at the end of the storage, there was a decrease tendency, indicating increase of intensity of blue color. As for T2, the value of  $b^*$  from the last day of storage was higher than the value at day 0, tending to a yellowish color stronger than the initial one. The increase in the values of  $a^*$ , as well as reduction in the values of  $b^*$  may indicate an oxidative browning.

The lowest reduction in the values of  $b^*$  was observed using coatings T3 (9.00%), T4 (11.06%) and control sample (16%). Treatment T2 showed an increase in the value of  $b^*$  of 2.20% in the last day of storage. Similar behavior was observed by Chagas *et al.* (2008), as the values of  $b^*$  decreased in 9 days of storage of minimally processed peaches treated with citric acid.

Days	Treatments			
	T1	T2	T3	T4
0	8.01 ± 0.92 <sup>eAB</sup>	7.02 ± 0.09 <sup>eB</sup>	8.64 ± 0.21 <sup>fA</sup>	7.52 ± 0.07 <sup>eAB</sup>
1	11.89 ± 0.31 <sup>cdA</sup>	7.93 ± 0.12 <sup>deD</sup>	9.44 ± 0.06 <sup>eB</sup>	8.54 ± 0.13 <sup>dc</sup>
3	12.87 ± 0.44 <sup>bcA</sup>	9.18 ± 0.35 <sup>cdB</sup>	12.29 ± 0.06 <sup>cdA</sup>	7.21 ± 0.40 <sup>eC</sup>
5	11.64 ± 0.08 <sup>dB</sup>	9.85 ± 0.29 <sup>cC</sup>	13.27 ± 0.18 <sup>bA</sup>	11.42 ± 0.11 <sup>aB</sup>
7	13.81 ± 0.15 <sup>bA</sup>	8.74 ± 0.08 <sup>cdD</sup>	12.14 ± 0.28 <sup>dB</sup>	10.85 ± 0.17 <sup>bC</sup>
9	13.70 ± 0.12 <sup>bA</sup>	12.42 ± 0.04 <sup>bb</sup>	12.63 ± 0.21 <sup>cb</sup>	10.23 ± 0.06 <sup>cc</sup>
12	15.49 ± 0.17 <sup>aAB</sup>	16.02 ± 0.06 <sup>aA</sup>	14.05 ± 0.05 <sup>bB</sup>	11.93 ± 0.16 <sup>ac</sup>

Means followed by the same letter in the column and capital letter in the line did not differ by Tukey's test ( $P < 0.05$ ). (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.

**TABLE 4.** VALUES OF  $a^*$  OF MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS, STORED AT  $4 \pm 1$  C FOR 12 DAYS

**TABLE 5.** VALUES OF  $b^*$  OF MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS, STORED AT  $4 \pm 1$  C FOR 12 DAYS

Days	Treatments			
	T1	T2	T3	T4
0	58.45 $\pm$ 0.38 <sup>AA</sup>	43.97 $\pm$ 0.21 <sup>EC</sup>	51.02 $\pm$ 0.24 <sup>BB</sup>	51.01 $\pm$ 0.06 <sup>BB</sup>
1	52.25 $\pm$ 1.35 <sup>BB</sup>	48.35 $\pm$ 0.17 <sup>AD</sup>	50.3 $\pm$ 0.13 <sup>CC</sup>	54.20 $\pm$ 0.18 <sup>AA</sup>
3	50.37 $\pm$ 0.93 <sup>CB</sup>	47.01 $\pm$ 0.16 <sup>BC</sup>	52.98 $\pm$ 0.31 <sup>AA</sup>	45.85 $\pm$ 0.23 <sup>DC</sup>
5	50.02 $\pm$ 0.07 <sup>CB</sup>	46.02 $\pm$ 0.20 <sup>CC</sup>	51.47 $\pm$ 0.22 <sup>BA</sup>	46.48 $\pm$ 0.19 <sup>CC</sup>
7	47.70 $\pm$ 0.26 <sup>DB</sup>	42.85 $\pm$ 0.12 <sup>FD</sup>	49.82 $\pm$ 0.21 <sup>CA</sup>	45.11 $\pm$ 0.26 <sup>EC</sup>
9	45.70 $\pm$ 0.13 <sup>EB</sup>	45.72 $\pm$ 0.16 <sup>CB</sup>	46.58 $\pm$ 0.25 <sup>DA</sup>	45.13 $\pm$ 0.21 <sup>EC</sup>
12	43.71 $\pm$ 0.21 <sup>FC</sup>	44.96 $\pm$ 0.13 <sup>DB</sup>	45.79 $\pm$ 0.32 <sup>EA</sup>	45.37 $\pm$ 0.19 <sup>DEAB</sup>

Means followed by the same letter in the column and capital letter in the line did not differ by Tukey's test ( $P < 0.05$ ). (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.

Table 6 shows values of Hue Angle ( $H^*$ ) of minimally processed peaches using different coatings stored at  $4 \pm 1$  C for 12 days.

There was a tendency to decrease the values of Hue Angle in all treatments applied to minimally processed peaches. This indicates that the tone changed from yellow to a reddish color.

The lowest percentage reduction was obtained in the treatment with tara gum (T4 – 7.8%), followed by xanthan gum (T3 – 9.3%) and sodium alginate (T2 – 13%). Uncoated treatment T1 showed reduction in the value of Hue Angle since the beginning of the experiment until the last day of storage, presenting the highest percentage reduction (15.6%) in the value of Hue Angle.

Similar behavior was observed by Martins (2010) with minimally processed peaches submitted to treatments with ascorbic acid and calcium chloride, as well as with different packaging materials. Values of Angle Hue decreased during storage from yellow-green tone to bright yellow, regardless of the treatment applied.

In the study by Costa (2010), a fluctuation in the values of Angle Hue during storage was observed. As for lightness, the lowest values were observed in the control treatment. On the other hand, the highest values (angles close to 90°)

were observed in the treatment with ascorbic and citric acid, which represents a more intense yellow.

Table 7 shows pH values of minimally processed peaches using different coatings, stored at  $4 \pm 1$  C for 12 days.

The pH of the control sample (T1) was maintained during storage, and at time zero it was significantly higher than the other treatments, as it had no addition of citric and ascorbic acids. There was a pH increase in the other treatments from the third and fifth day of storage. However, there was no significant difference among treatments at the end of storage. Maintenance or even increasing of pH values is expected when calcium chloride is added, as this compound is a salt chlorinate of basic nature providing buffer capacity (Andrade 2006).

Different results have been reported in literature. Chagas *et al.* (2008) showed that in minimally processed peaches treated with different concentrations of citric acid the pH was maintained throughout the storage period. At the end of storage there was no difference between the treatments with citric acid. Studies performed by Togrul and Arslan (2004) and Maftoonazad *et al.* (2008) with coating of whole peaches showed similar results to the ones obtained in the present work as there was increase in the pH during storage.

**TABLE 6.** VALUES OF HUE ANGLE ( $H^*$ ) OF MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS STORED AT  $4 \pm 1$  C FOR 12 DAYS

Days	Treatments			
	T1	T2	T3	T4
0	82.2 $\pm$ 0.16 <sup>AA</sup>	80.93 $\pm$ 0.11 <sup>AC</sup>	80.39 $\pm$ 0.27 <sup>AD</sup>	81.61 $\pm$ 0.07 <sup>AB</sup>
1	77.18 $\pm$ 0.51 <sup>BC</sup>	80.73 $\pm$ 0.07 <sup>AA</sup>	79.37 $\pm$ 0.04 <sup>BB</sup>	81.08 $\pm$ 0.16 <sup>AA</sup>
3	75.64 $\pm$ 0.78 <sup>CD</sup>	78.92 $\pm$ 0.11 <sup>BB</sup>	76.94 $\pm$ 0.37 <sup>CC</sup>	81.06 $\pm$ 0.33 <sup>AA</sup>
5	75.49 $\pm$ 0.13 <sup>CC</sup>	78.01 $\pm$ 0.19 <sup>CA</sup>	75.54 $\pm$ 0.15 <sup>DECD</sup>	76.2 $\pm$ 0.15 <sup>CB</sup>
7	73.95 $\pm$ 0.11 <sup>DC</sup>	78.55 $\pm$ 0.03 <sup>BCA</sup>	76.32 $\pm$ 0.20 <sup>CD</sup>	76.47 $\pm$ 0.23 <sup>CB</sup>
9	73.38 $\pm$ 0.11 <sup>DC</sup>	74.92 $\pm$ 0.46 <sup>DB</sup>	74.82 $\pm$ 0.17 <sup>EB</sup>	77.23 $\pm$ 0.16 <sup>BA</sup>
12	69.41 $\pm$ 0.16 <sup>ED</sup>	70.39 $\pm$ 0.12 <sup>EC</sup>	72.94 $\pm$ 0.09 <sup>FB</sup>	75.26 $\pm$ 0.18 <sup>DA</sup>

Means followed by the same letter in the column and capital letter in the line did not differ by Tukey's test ( $P < 0.05$ ). (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.

Days	Treatments			
	T1	T2	T3	T4
0	4.01 ± 0.02 <sup>Aa</sup>	3.79 ± 0.03 <sup>cB</sup>	3.78 ± 0.05 <sup>Bbc</sup>	3.70 ± 0.03 <sup>bC</sup>
1	4.01 ± 0.01 <sup>aA</sup>	3.80 ± 0.02 <sup>bcB</sup>	3.78 ± 0.02 <sup>bb</sup>	3.71 ± 0.01 <sup>bC</sup>
3	4.09 ± 0.11 <sup>aA</sup>	3.86 ± 0.08 <sup>bcB</sup>	3.98 ± 0.03 <sup>aAB</sup>	3.95 ± 0.04 <sup>aAB</sup>
5	4.11 ± 0.05 <sup>aA</sup>	4.00 ± 0.07 <sup>abA</sup>	4.10 ± 0.09 <sup>aA</sup>	4.04 ± 0.07 <sup>aA</sup>
7	4.17 ± 0.09 <sup>aA</sup>	4.13 ± 0.10 <sup>aAB</sup>	4.05 ± 0.05 <sup>aAB</sup>	3.96 ± 0.02 <sup>aB</sup>
9	4.24 ± 0.03 <sup>aA</sup>	4.20 ± 0.09 <sup>aA</sup>	4.02 ± 0.02 <sup>aB</sup>	3.95 ± 0.03 <sup>aB</sup>
12	4.02 ± 0.25 <sup>aA</sup>	4.14 ± 0.08 <sup>aA</sup>	4.01 ± 0.01 <sup>aA</sup>	3.92 ± 0.09 <sup>aA</sup>

Means followed by the same letter in the column and capital letter in the line did not differ by Tukey's test ( $P < 0.05$ ). (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.

Table 8 shows values of content of total soluble solids (°Brix) of minimally processed peaches using different coatings, stored at 4 ± 1°C for 12 days.

The content of soluble solids was influenced by the treatment applied to minimally processed peaches. In treatments T1 and T2, there was an increase in soluble solids from the first and the last day of storage, but in treatments T3 and T4, there was a reduction in soluble solids.

The increase in the content of total soluble solids observed in control sample (T1) and in the treatment with alginate (T2) can be the result of sugars accumulation, which is concentrated by loss of moisture, a process that occurs during fruit ripening even though in small scale. This ripening was restricted in the treatments with xanthan gum (T3) and tara gum (T4). The reduction of total soluble solids content observed in T3 and T4 can be attributed to the fact that the immersion of fruits in filmogenic solutions may have leached the total soluble solids of the fruit (Trigo 2010). Tendency of increase in the values of pH of minimally processed peaches was observed by Costa (2010). In their study, both the control sample and the ones treated with ascorbic acid and calcium chloride; L-cysteine hydrochloride and calcium chloride; L-cysteine hydrochloride, ascorbic acid and calcium chloride, showed this tendency.

Days	Treatments			
	T1	T2	T3	T4
0	12.67 ± 0.07 <sup>bcA</sup>	10.3 ± 0.10 <sup>eC</sup>	10.83 ± 0.02 <sup>aB</sup>	12.6 ± 0.02 <sup>aA</sup>
1	12.60 ± 0.08 <sup>cA</sup>	10.4 ± 0.09 <sup>deC</sup>	10.90 ± 0.05 <sup>aB</sup>	12.63 ± 0.05 <sup>aA</sup>
3	11.90 ± 0.05 <sup>dA</sup>	10.6 ± 0.02 <sup>dC</sup>	10.90 ± 0.07 <sup>aC</sup>	11.30 ± 0.11 <sup>bcB</sup>
5	11.97 ± 0.09 <sup>dA</sup>	11.0 ± 0.01 <sup>cB</sup>	10.90 ± 0.05 <sup>aB</sup>	10.9 ± 0.13 <sup>cB</sup>
7	12.60 ± 0.10 <sup>cA</sup>	12.0 ± 0.01 <sup>aB</sup>	10.90 ± 0.01 <sup>aD</sup>	11.10 ± 0.10 <sup>bcC</sup>
9	12.90 ± 0.11 <sup>abA</sup>	11.73 ± 0.09 <sup>abB</sup>	10.3 ± 0.12 <sup>bc</sup>	11.30 ± 0.30 <sup>bcB</sup>
12	13.10 ± 0.2 <sup>aA</sup>	11.63 ± 0.07 <sup>bb</sup>	9.4 ± 0.17 <sup>cC</sup>	11.60 ± 0.30 <sup>bb</sup>

Means followed by the same letter in the column and capital letter in the line did not differ by Tukey's test ( $P < 0.05$ ). (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.

**TABLE 7.** pH VALUES OF MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS STORED AT 4 ± 1°C FOR 12 DAYS

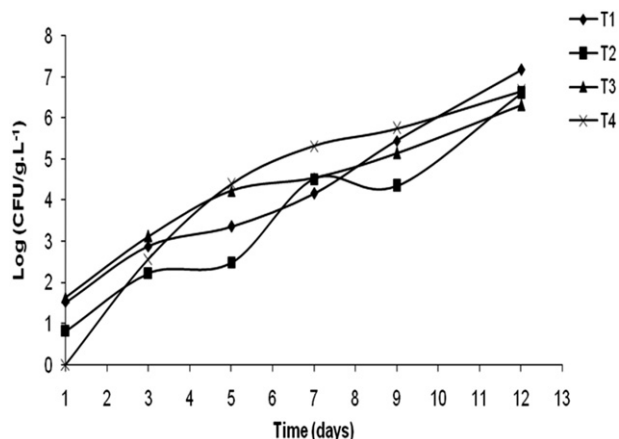
Martins (2010) observed that soluble solids content (°Brix) remained constant during the storage of minimally processed peaches, with no significant difference of the values among different treatments with ascorbic acid and calcium chloride. Nevertheless, when different packages were evaluated, different results were observed. The highest values of soluble solids were found in products from treatments with packaging PD 955 and PD 900 (polyolefin); according to the authors, this behavior is a result of biochemical reactions resulting from the ripening. PET, PVC 11 µm and PVC 14 µm packages provided a higher restraint of the ripening of products during storage. Chagas *et al.* (2008) also did not observe differences in the content of soluble solids when minimally processed peaches were treated with 1% and 2% citric acid and stored for 9 days at 5°C.

### Microbiological Analyses

Figure 1 shows growth curves of psychotrophic microorganisms in minimally processed peaches with and without coating, stored for 12 days.

In the analysis of psychotrophic species, the initial count for all treatments was below 2 log cfu/g, and treatment 4

**TABLE 8.** CONTENT OF TOTAL SOLUBLE SOLIDS (°BRIX) OF MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS STORED AT 4 ± 1°C FOR 12 DAYS



**FIG. 1.** COUNTING OF PSYCHOTROPIC MICROORGANISMS IN MINIMALLY PROCESSED PEACHES USING DIFFERENT COATINGS (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.

(with tara gum) was the one that had the lowest initial growth for these microorganisms. Throughout the storage period, there was an increase in the count and a similar behavior in relation to psychotropic growth. However, growth of these microorganisms in the control sample (T1) was higher than the growth with coatings in the 12 days of storage, reaching 7.17 log cfu/g. However, in treatments with coatings, the final count was 6.6 log cfu/g for treatment 2, 6.31 log cfu/g for treatment 3 and 6.65 log cfu/g for treatment 4.

Figure 2 shows growth curves of molds and yeasts for all treatments applied to minimally processed peaches.

As it can be seen in Fig. 2, treatments T3 and T4 had similar behavior for the growth of molds and yeasts and these treatments presented lower growth for these microorganisms, when compared with the other treatments. At the end of the storage period, these treatments reached 4.0 log cfu/g and 3.9 log cfu/g, respectively. The highest count of molds and yeasts occurred in the control sample (5.49 log cfu/g) followed by treatment T2 (5.12 log cfu/g). There are no reports in literature about microbiological evaluation of psychotropic microorganisms, molds and yeasts in minimally processed peaches.

According to Kester and Fennema (1986), the use of coatings with low gas permeability, as it is the case of polysaccharides, reduces the access to oxygen, minimizing microbiological alterations. Similar behavior to the one observed in this study was verified by Jacometti *et al.* (2003) in whole peaches coated with gelan gum and stored at 10C. After 2 weeks of storage, coated peaches presented lower counts of psychotropic microorganisms than uncoated

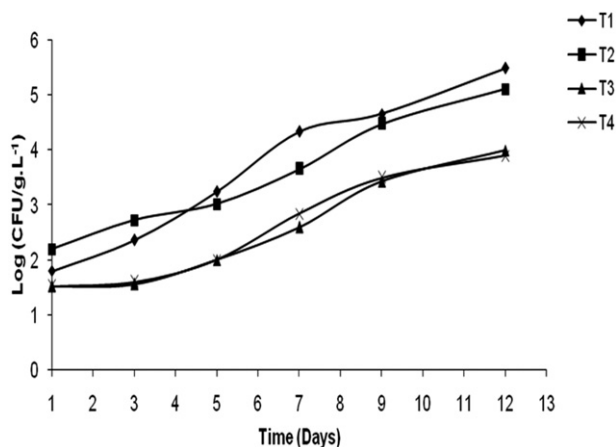
fruits; however, there was no difference in mold and yeast counts between coated samples and control.

Olivas *et al.* (2007) in their studies with minimally processed apples coated with alginate and stored at 4C, found low levels of psychotrophs (10<sup>1</sup> cfu/g) and molds and yeasts (<10<sup>1</sup> log cfu/g) during the whole period of storage. These counts were lower than the ones found in the present study in treatment 2 with alginate in which 10<sup>6</sup> cfu/g were obtained for psychotrophs and 10<sup>5</sup> cfu/g for molds and yeasts.

It has been recognized that foods with microbial counts above 10<sup>6</sup> cfu/g are unsuitable for human consumption, due to loss of nutritional value, sensorial alterations and contamination risks (Vitti *et al.* 2004). Therefore, all samples analyzed became unsuitable for consumption after 9 days of storage, due to psychotrophic microorganisms. Regarding mold and yeast counts, in 12 days, all samples were still below the limit.

## CONCLUSION

It can be concluded that the different coatings used in this work were effective in the preservation of minimally processed peaches, when compared with the control sample. Regarding the coatings evaluated, treatment with tara gum associated to citric and ascorbic acids and to calcium chloride showed the best results, especially when compared to alginate treatment, which is the main polysaccharide currently used as edible coating. Tara gum caused reduction in mass loss, maintaining firmness, less color alteration and lower growth of yeasts and molds. Thus, this gum has



**FIG. 2.** GROWTH OF MOLDS AND YEASTS IN MINIMALLY PROCESSED PEACHES WITH DIFFERENT COATINGS (T1) control; (T2) alginate 2%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T3) xanthan gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%; (T4) tara gum 0.5%, ascorbic acid 1%, citric acid 0.25%, CaCl<sub>2</sub> 1%, glycerol 1%.



potential for the application as coating in minimally processed peaches from Granada cultivar.

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